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# 1-Chloro-1,1-difluoro-N-(4-methoxyphenyl)-3-(pyrrolidin-2-ylidene)propan-2-imine

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In the title compound,  $C_{14}H_{15}ClF_2N_2O$ , the Z configuration has been confirmed. The molecular structure shows an intramolecular N-H···N hydrogen bond [H···N 2.04 (6), N···N 2.709 (6) Å and N–H···N 124 (5)°]. This interaction could be responsible for the Z configuration.

## Comment

Fluorine-containing amino compounds have found a wide range of synthetic and biochemical applications due to the unique biological properties imparted by fluorine (Soloshonok et al., 1997, 1998; Bégué & Bonnet-Delpon, 1996). Due to the increasing interest in the synthesis and reactivity of 1,3difunctionalized fluorine-containing organic compounds (Uneyama et al., 1993; Fustero et al., 1996), we have focused our attention on the fluorinated vinylogous amidines.

We have recently developed the reaction of fluorinated imidoyl chlorides with ketimines to afford the first reported fluorinated vinylogous amidines (Fustero et al., 1999). Although, several tautomeric  $\alpha - \gamma$  structures were possible, the  $\beta$  enaminic structure was determined by NMR analyses. However, the sterochemical assignment (Z or E) was not fully confirmed. Two-dimensional HOESY NMR <sup>1</sup>H-<sup>19</sup>F experiments suggest the Z configuration for the  $\beta$  enamino tautomer (Fustero et al., 1999). We have now determined the first crystal structure of a fluorinated vinylogous amidine which confirms the assigned geometry.

In the molecular structure of the title compound, (I), the pyrrolidine ring is planar (mean deviation 0.028 Å), with the N1-C1 bond length [1.326(7) Å] shorter than N1-C4 [1.451 (7) Å]. This suggests an electronic elocalization along the imino-enamino group with bond lengths C1-C5, C5-C6 and C6-N2 of 1.380 (8), 1.417 (7) and 1.312 (7) Å, respectively. The pyrrolidine ring NH group acts as a hydrogen-bond donor to the iminic N atom in the same molecule. The resulting intramolecular hydrogen bond N1-H1···N2  $[H1 \cdots N2 \ 2.04 \ (6), \ N1 \cdots N2 \ 2.709 \ (6) \ \text{Å} and \ N1 - H1 \cdots N2$  $124(5)^{\circ}$  makes a planar six-membered ring N2-C6-C5-C1-N1-H1 (mean deviation 0.014 Å). The pyrrolidine ring and the hydrogen-bonded ring adopt a nearly coplanar conformation (dihedral angle  $3.6^{\circ}$ ); this plane is almost perpendicular to the aromatic ring  $(83.5^{\circ})$ .



# **Experimental**

*n*-Butyllithium (2.5 *M* in hexane, 2.8 ml, 7 mmol) was added dropwise to a solution of diisopropylamine (1 ml, 7 mmol) in THF (10 ml) at 273 K under argon. After being stirred for 10 min, 2-methyl-1pyrroline (3.5 mmol) in THF (7 ml) was added. The resulting mixture was stirred at 273 K for 30 min, and then cooled to 195 K. N-(p-Anisidino)-2-chloro-2,2-difluoroacetimidoyl chloride (3.5 mmol) in THF (7 ml) was added and stirred at 195 K. After 1 h, the reaction was quenched by addition of saturated ammonium chloride solution. The aqueous layer was extracted with dichloromethane  $(3 \times 25 \text{ ml})$ . The combined organic extracts were washed with brine, then dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated under reduced pressure to furnish the crude product. Flash chromatography [n-hexane-EtOAc (3:1)] on silica gel gave a yellow solid (79%, m.p. 361-363 K). Single crystals of (I) suitable for X-ray diffraction were grown by cooling a hexane solution.

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$C_{14}H_{15}ClF_2N_2O$	$D_x = 1.426 \text{ Mg m}^{-3}$
$M_r = 300.73$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 25
$a = 7.6548 (15) \text{\AA}$	reflections
b = 10.436(2) Å	$\theta = 9.5 - 12.2^{\circ}$
c = 17.543 (4)  Å	$\mu = 0.292 \text{ mm}^{-1}$
$\beta = 91.34 \ (3)^{\circ}$	T = 293 (2)  K
$V = 1401.0 (5) \text{ Å}^3$	Prism, yellow
Z = 4	$0.78 \times 0.46 \times 0.42 \text{ mm}$

Data collection

Enraf-Nonius CAD-4 diffractometer  $\omega$  scans 3826 measured reflections 2453 independent reflections 1774 reflections with  $I > 2\sigma(I)$  $R_{\rm int} = 0.020$ 

### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.070$  $wR(F^2) = 0.208$ S = 1.0382453 reflections 185 parameters H atoms treated by a mixture of independent and constrained refinement

 $\theta_{\rm max} = 24.97^{\circ}$  $h = -9 \rightarrow 9$  $k = -12 \rightarrow 5$  $l=0\rightarrow 20$ 3 standard reflections every 247 reflections intensity decay: 1.6%

 $w = 1/[\sigma^2(F_o^2) + (0.1017P)^2]$ + 1.8583P] where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.59 \ {\rm e} \ {\rm \AA}^{-3}$  $\Delta \rho_{\rm min} = -0.65 \text{ e } \text{\AA}^{-3}$ 

#### Table 1

Selected geometric parameters (Å, °).

N1-C1	1.326 (7)	C1-C2	1.457 (7)
N1-C4	1.451 (7)	C2-C3	1.429 (8)
N2-C6	1.312 (7)	C3-C4	1.537 (8)
N2-C11	1.419 (6)	C5-C6	1.417 (7)
C1-C5	1.380 (8)		
C1-N1-C4	113.6 (4)	C2-C3-C4	105.5 (5)
C6-N2-C11	126.1 (4)	N1-C4-C3	103.1 (5)
N1-C1-C5	123.9 (4)	C1-C5-C6	124.6 (5)
N1-C1-C2	108.2 (5)	N2-C6-C5	121.5 (5)
C5-C1-C2	127.6 (5)	N2-C6-C7	122.0 (4)
C3-C2-C1	109.1 (5)	C5-C6-C7	116.5 (4)

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$N1 - H1 \cdots N2$	0.97 (4)	2.04 (6)	2.709 (6)	124 (5)

The amino-group H atom was found in a difference Fourier synthesis and refined with a restrained N—H bond length. Other H atoms were refined using a riding model (C—H 0.93-0.98 Å).

Data collection: *CAD*-4–96 *Software* (Enraf–Nonius, 1996); cell refinement: *CAD*-4–96 *Software*; data reduction: *MolEN* (Fair, 1989); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997);

program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); software used to prepare material for publication: *SHELXL*97.

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#### References

- Bégué. J. P. & Bonnet-Delpon, D. (1996). Bimoedical Frontiers of Fluorine Chemistry, edited by I. Ojima, J. R. McCarthy & J. Welch, pp. 59–72. Washington, DC: American Chemical Society Books.
- Enraf-Nonius (1996). CAD-4-96 Software. Enraf-Nonius, Delft, The Netherlands.
- Fair, C. K. (1989). MolEN. Enraf-Nonius, Delft, The Netherlands.
- Fustero, S., de la Torre, M. G., Pina, B. & Simón Fuentes, A. (1999). J. Org. Chem. 64, 5551–5556.
- Fustero, S., Navarro, A., Díaz, D., de la Torre, M. G., Asensio, A., Sanz, F. & Liu, M. G. (1996). J. Org. Chem. 61, 8849–8859.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Soloshonok, V. A., Ono, T. & Soloshonok, I. V. (1997). J. Org. Chem. 62, 7538– 7539.
- Soloshonok, V. A., Soloshonok, I. V., Kuhkar, V. P. & Svedas, V. K. (1998). J. Org. Chem. 63, 1878–1884.
- Uneyama, K., Tamura, K., Mizukami, H., Maeda, K. & Watanabe, H. (1993). J. Org. Chem. 58, 32–35.